

Journal of Scientific Research & Reports

27(10): 26-30, 2021; Article no.JSRR.73946 ISSN: 2320-0227

# Natural Radioactivity in Various Water Tasters Using $\gamma$ -Ray Measurement

I. Hossain<sup>1\*</sup>, N. M. Yussuf<sup>2</sup>, M. A. Saeed<sup>3</sup>, M. O. Alzanbaqi<sup>1</sup> and H. Wagiran<sup>2</sup>

<sup>1</sup>Department of Physics, Rabigh College of Sciences and Arts, King Abdulaziz University, Rabigh 21911, Saudi Arabia. <sup>2</sup>Department of Physics, Faculty of Science, Universiti Teknologi Malaysia, 81310 Skudai, Johor, Malaysia. <sup>3</sup>Department of Physics, Division of Science & Technology, University of Education, Township, Lahore, Pakistan.

## Authors' contributions

This work was carried out in collaboration among all authors. All authors read and approved the final manuscript.

## Article Information

DOI: 10.9734/JSRR/2021/v27i1030446 <u>Editor(s):</u> (1) Lesław Juszczak University of Agriculture in Krakow, Poland. <u>Reviewers:</u> (1) Munkhsaikhan Gonchigsuren, Mongolian University of Science and Technology, Mongolia. (2) Hasratuddin, Medan State University, Indonesia. Complete Peer review History: <u>https://www.sdiarticle4.com/review-history/73946</u>

Original Research Article

Received 12 July 2021 Accepted 20 September 2021 Published 29 October 2021

# ABSTRACT

This paper has explained the contamination of natural radionuclides in various water testers using gamma ray measurement which is very significant as part of health scrutiny programs to progress the ecological knowledge. Natural radioactivity was determined in five groups of water samples (rain, mine, tap, drinking and mineral) from different places at Johor, Malaysia by means of gamma-ray spectrometry tool. The annual cumulative effective doses were estimated 6.05 mSv /yr for rain, 9.49 mSv/ yr for mine, 6.39 mSv /yr for tap, 5.67 mSv /yr for drinking, and 6.01 mSv/yr for mineral water. Among the five samples, mine water gave the highest value in annual effective dose measurement. The measured data are compared with the reported value. The activity concentrations of five water samples provided that bottled drinking water was the lowest than other water samples. This research is useful to provide some information to the public about the amount of radionuclide content uranium, thorium and potassium that present in water.

\*Corresponding author: E-mail: mihossain@kau.edu.sa, hossain196977@yahoo.com;

Keywords: Water; Malaysia; potassium; thorium; uranium.

## **1. INTRODUCTION**

During the last decades, the global pollution by radioisotopes and after that, nuclear blasts in higher layers of the atmosphere polluted the earth [1-2] callously. The dangerous radionuclides, formed owing to individual's explosions, polluted the complete environment such as layers of soil, food and water etc. These hazardous isotopes go to the human body by taking drinking water and food. Fresh intake water is very important to human and natural life. Safe drinking water has developed gradually and significantly almost everywhere in the world [3-4].

Radio-isotopes go inside the body and work together with interior tissue. Radium and Uranium isotopes are investigated in intake water and they don't produce great quantities of gamma radiation, so showering and washing do not carry major risk. Nevertheless, these radionuclides are consumed through eating or drinking; the radiations can absorb directly with the body and interact with tissues. Uranium is a naturally-occurring radionuclide which can melt with soils and minerals with water, subsequently at high levels of radio-isotopes can exist in groundwater. Uranium has a destructive effect on the kidney during long-term drinking water and clues to cancer [5].

Fatima et al [6] and Ben et al. [7] studied radionuclide of gamma-ray emitters in bottled water. Kawabata et al. [8] and Somlaj et al. [9] measured radioactivity of Uranium and thorium in soil and water. Salih et al. [10] measured natural radioactivity <sup>238</sup>U, <sup>226</sup>Ra, <sup>228</sup>Ra, <sup>222</sup>Rn and <sup>210</sup>Po in drinking groundwater samples. Para et al. [11] and Rahman et al. [12] determined radioactivity levels of U, Th and other impurities in Molybdenum by radiochemical neutron activation analysis. Yusuf et al. [13] determined usual radiation in mineral and drinking water in Johr Bahru, Malaysia. In this paper, we would like to perform the existence of usual radio-nuclei in different tasters of water usina γ-rav measurement.

# 2. STATEMENT OF PROBLEM

Water has taken in the body in a variety of ways. They are vital to our health and well-being. The existence of radionuclides in drinking water causes health threats owing to human interior exposure from the decay of radionuclides immersed into the body through ingestion. Many

countries have conducted studies to resolve this problem. Radiation exposure due to water will be directly to the public through the internal and external exposure. When the gamma ray emits inside and outside the body, it will contribute to the collective's dose in individuals. The purpose of this work for monitoring of normal radioactivity in drinking water is a significant parameter for public health studies, which permits the assessment of population exposure to radiation by the ingesting of water. The measure of concentration of the radionuclides can be calculated to find the effective dose per individual and rising awareness over the polluted drinking water correlated to sicknesses.

This research measured the amount of natural radionuclides in different sources of water that are usually taken in Malaysia and also for many countries. The results obtained from this research were compared with reported data and from the internationally recommended values as sated in ICRP (International Commission Radiation Protection). Since there are growing concerns about the amount of radioactive source, this research is useful to provide some information to the public about the amount of radionuclide content which are uranium, thorium and potassium that present in water. It is important to know the radionuclide contained in each source of water so that we can optimize the dose level from the human risk from overexposure. This research shows only incremental advance over prior research results that display the originality of this work.

## 3. EXPERIMENTAL TOOLS

# 3.1 Sampling

The water tasters were taken from different locations in Malaysia. The first water sample is rainwater that falls in the middle of the city of Johor Bahru. The second water sample was taken from mining areas in Taiping, Perak. The third sample is tap water from residential areas surrounded by an industrial area in Pasir Gudang, Johor. The fourth and fifth samples are bottled drinking and mineral water sold in stores around Johor Bahru. Each sample of water about one liter was kept in bottles and the pure PH data was determined and extra nitric acid (HNO<sub>3</sub>) was added until the PH analysis shows 1-3. The HNO<sub>3</sub> was useful to hold the component of water from disappearing or lacking. The water evaporated until 0.5 liters. In order to find

equipoise for  $\gamma$ -ray spectroscopy, the samples were taken in Marinelli cup for a month.

#### 3.2 Instrumentation

The measurements were done using facilities of  $\gamma$ -ray spectroscopy tool at the Department of Universiti Teknologi Physics, Malaysia. It comprises a high purity germanium (HPGe) detector, pre-amplifier, amplifier, scalar and multi-channel analyzer (MCA). The efficiency of the detector is 0.2. The resolution of the detector is 1.8 x 10<sup>3</sup> eV at peak energy 1.333 MeV. To protect the detector from cosmic rays, lead shields of 470 mm thick, 2 mm coated with tin and copper were used. The <sup>137</sup>Cs, <sup>60</sup>Co, <sup>152</sup>Eu etc sources were used for energy calibrations of the gamma ray spectra.

#### 4. RESULTS AND DISCUSSION

The concentration of  $U^{238}$ ,  $Th^{232}$ , and  $K^{40}$  are obtained using equation (1).

$$C_s = \frac{N_s C_p}{N_p} \tag{1}$$

Where,  $N_p$  counts of standard,  $N_s$  are counts of sample,  $C_s$  concentration of sample.  $C_p$  concentration of standard and the uncertainty or error ( $\Delta$ ) of concentration are found using equation (2)

$$\Delta C_{s} = \left(\frac{\Delta Ns}{Ns} + \frac{\Delta Cp}{Cp} + \frac{\Delta NP}{NP}\right)$$
(2)

Where,  $N_s$  stands for sample counts,  $N_P$  for standard counts,  $\Delta C_s$  is error for concentration,  $\Delta N_s$  error for tester counts,  $\Delta N_P$  error standard counts, *NP* indicate mass standard,  $\Delta NP$  indicate uncertainty mass standard.

Table 1 shows specific activity of U<sup>238</sup>, Th<sup>232</sup> and K<sup>40</sup> in 5 water testers. The activity of U<sup>238</sup>, Th<sup>232</sup> and K<sup>40</sup> in the five water testers ranged from 18.20  $\pm$  3.20 Bq /Kg to 40.83  $\pm$  1.23 Bq/ kg, 38.68  $\pm$  0.44 Bq/ kg to 110.10  $\pm$  0.32 Bq/ kg and 122.19  $\pm$  0.01 Bq /Kg to 201.06  $\pm$  0.01 Bq/Kg respectively. Mine water samples had the highest specific activity of all radionuclide's (U<sup>238</sup>, Th<sup>232</sup> and K<sup>40</sup>) while drinking water samples had the lowest activity.

The specific activity of standard soil samples, SL-1 supplied by IAEA for different radio-nuclides of  $U^{238}$ ,  $Th^{232}$  and  $K^{40}$  were tabulated in Table 2. Comparison of mean and range of concentration of  $U^{238}$ ,  $Th^{232}$  and  $K^{40}$  in present work with those

of ref. [14-18] are also presented in Table 2. The mean activity was 28.24 ± 8.13, 65.79 ± 26.62 and 143.29 ± 32.70 Bg/ kg for U<sup>238</sup>, Th<sup>232</sup> and K<sup>40</sup> respectively in the present work. Thus <sup>40</sup>K is the major activity and <sup>238</sup>U is the smallest activity in the water testers of present work. Desideri et al., [14, 16] determined the mean and range concentration of U<sup>238</sup> of mineral waters in Italy. The values are 8.97  $\pm$  14.72 mBqL<sup>-1</sup> and 0.17 to 89.00 mBqL<sup>-1</sup> and 9.52±10.92 mBqL<sup>-1</sup> and 0.65-48.77 mBqL<sup>-1</sup> respectively. Ajayi et al. [15] measured 19.09 ± 10.05 Bgkg<sup>-1</sup> and 0.57±0.21-34.08±5.61 Bq/kg mean and range activity concentration of K<sup>40</sup> in drinking water in Nigeria. Fatima et al. [17] determined the mean and range concentration of Th<sup>232</sup> and K<sup>40</sup> in bottled drinking water in Pakistan 5.2 ± 0.9 mBqL<sup>-1</sup> and 4  $\pm 0.5 - 6 \pm 0.8$  mBgL<sup>-1</sup> and 140.9  $\pm 30.6$  mBgL<sup>-1</sup> and 92  $\pm$  5 - 216  $\pm$  10 mBqL<sup>-1</sup> respectively. Ajayi et al. [18] measured 2.94 BqL<sup>-1</sup> and 1.74 – 4.69 BqL<sup>-1</sup> for mean and range activity concentration in drilled and dug well drinking water of Ogun state southwestern Nigeria.

The results of the calculated yearly actual dose  $(\mu Sv/yr)$  are given in Table 3. The yearly actual dose owing to the consumption of <sup>238</sup>U, <sup>232</sup>Th and  $^{40}$ K from water testers are 0.01 ± 0.007 to 0.03 ± 0.01  $\mu$ Sv yr<sup>-1</sup>, 0.12 ± 0.06 to 0.35 ± 0.08  $\mu$ Sv yr<sup>-1</sup> and 5.54  $\pm$  1.38 to 9.91 $\pm$  1.48  $\mu$ Sv yr<sup>-1</sup> respectively. The mean yearly actual doses from  $^{238}$ U to the age group is 0.02 ± 0.007 µSv /yr and that due to  $^{232}$ Th in the sample is 0.21 ± 0.08  $\mu$ Sv/yr and also the mean annual effective dose for  ${}^{40}$ K is 6.49 ± 1.48  $\mu$ Sv/yr. The yearly actual doses from <sup>40</sup>K to the age group was the highest than the <sup>232</sup>Th and <sup>238</sup>U. The total annual effective dose varied from 5.67  $\pm$  1.34 to 9.49  $\pm$ 1.57  $\mu$ Sv yr<sup>-1</sup> and with a mean of 6.72 ± 1.57  $\mu$ Sv yr<sup>-1</sup>. This study shows radio-logically safe to use the water testers. We found that all testers gave much lower internal exposures than the United Nation Scientific Committee on Effects of Atomic Radiation (UNSCEAR) [1] testified world average value of 0.12 mSvyr<sup>-1</sup> and the WHO [19] and ICRP [20] reference limits of 0.1 mSvyr<sup>-1</sup> respectively. The UNSCEAR [1] reported the average worldwide ingestion exposure (about 0.17 mSv/yr from K-40 and about 0.12 mSv/yr from U-238 and Th-232) for usual food and drinking water. The water testers of annual effective dose for  $^{238}$ U,  $^{232}$ Th and  $^{40}$ K were 0.02 ±  $0.007 \ \mu Svyr^1$ ,  $0.21 \pm 0.08 \ \mu Svyr^1$  and  $6.49 \pm 1.48 \ \mu S$ vyr<sup>1</sup> respectively. Therefore these testers have a greatly lower actual dose than the report of the UNSCEAR.

Samples	<sup>238</sup> U (Bq kg <sup>-1</sup> )	<sup>232</sup> Th (Bq kg <sup>-1</sup> )	<sup>40</sup> K (Bq kg⁻¹)
Rain	28.78±1.35	60.74±0.40	128.85±0.01
Mine	40.83±1.23	110.10±0.32	201.06±0.01
Тар	26.81±1.35	55.41±0.44	136.63±0.01
Drinking	18.20±3.20	38.68±0.44	122.19±0.01
Mineral	26.57±1.60	64.04±0.40	127.74±0.01

#### Table 1. Specific Activity, As

 Table 2. Comparison of mean and range of concentration of <sup>238</sup>U, <sup>232</sup>Th and <sup>40</sup>K in present work with those of ref. [14-18]

Samples	Annual Effective Dose (± 0.01µSv yr <sup>-1</sup> )				
	<sup>238</sup> U	<sup>232</sup> Th	<sup>40</sup> K	Total	
Rain	0.02	0.19	5.84	6.05	
Mine	0.03	0.35	9.11	9.49	
Тар	0.02	0.17	6.19	6.39	
Drinking	0.01	0.12	5.54	5.67	
Mineral	0.02	0.20	5.79	6.01	
Range	0.01 – 0.03	0.12 – 0.35	5.54 – 9.11	5.67 – 9.49	
Mean	0.02	0.21	6.49	6.72	
Standard Deviation	0.007	0.08	1.48	1.57	

#### **Table 3. Annual Effective Dose**

Radio -nucl.	Standard Con. ( Bq kg <sup>-1</sup> )	Present Mean ( Bq kg <sup>-1</sup> )	Ref. Mean ( Bq kg <sup>-1</sup> )	Present Range ( Bq kg <sup>-1</sup> )	Ref. Range (Bq kg⁻¹)
	58.06 ±	28.24 ± 8.13	0.009 ±	18.20 ± 8.13 to 40.83	0.0002 to
<sup>238</sup> U	0.12		0.001[14]	± 8.13	0.089[14]
			0.010 ±		0.0007 to
			0.011[16]		0.048[16]
	400.15.±0.	65.79 ±26.62	0.005 ±	38.68 ± 26.62 to	0.004 ±0.001 to
<sup>232</sup> Th	04		0.001[17]	110.10 ± 26.62	0.06 ± 0.001 [17]
	119.97	143.29 ±32.7	19.09 ±	122.19 ± 32.70 to	0.57 ± 0.21 to
<sup>40</sup> K	±0.03		10.05[15]	201.06 ± 32.70	34.08 ± 5.61[15]
			0.14 ±		0.09 ± 0.01 to
			0.03[17]		0.22 ± 0.01[17]
			2.94± 0.79[18]		1.74 to 4.69[18]

#### **5. CONCLUSION**

This ecological study was carried out to provide information about concentrations of radionuclides in five groups of water testers (rain, mine, tap, drinking and mineral) from different places at Johor, Malaysia by using high purity Ge type gamma spectrometers. The activity of U<sup>238</sup>, K<sup>40</sup>, and Th<sup>232</sup> (Bq/ Kg) were measured 28.78 ± 1.35, 128.85 ± 0.01 and 60.74 ± 0.40 respectively for rain water, 40.83 ± 1.23, 201.06 ± 0.01 and 110.10 ± 0.32 for mine water, 26.81 ± 1.35, 136.63 ± 0.01 and 55.41 ± 0.44 for tap water, 18.20 ± 3.20, 122.19 ± 0.01 and 38.68 ± 0.44 for drinking water, 26.57 ± 1.60, 127.74 ± 0.01 and 64.04 ± 0.40 for mineral water. The mean activity of five group of water was  $28.24 \pm 8.13$ ,  $65.79 \pm$ 26.62 and 143.29  $\pm$  32.70 Bq/ kg for U^{238}, Th^{232} and K<sup>40</sup> respectively in the present work. This study showed that mine water consist the highest specific activity in <sup>238</sup>U, <sup>232</sup>Th and <sup>40</sup>K which 40.83 ± 1.23 Bq/ kg, 110.10 ± 0.32 Bq/ kg and 201.06 ± 0.01 Bq/ kg respectively as well as high in volume of the annual effective dose which 9.49 ±  $0.01 \,\mu$ Sv /yr. The drinking water had the lowest activity of 18.20 ± 3.20 Bq/ kg in <sup>238</sup>U, 38.68 ± 0.44 Bg /kg in  $^{232}$ Th and 122.19 ± 0.01 Bg/kg in <sup>40</sup>K respectively. All the five water testers are safe compared to the UNSCEAR report. Present testers annual estimated effective doses were less compared to the World Health Organization (WHO) limit 0.1mSv yr<sup>-1</sup>.

#### ACKNOWLEDGEMENT

The author thanks to department of physics, Universiti Teknologi Malaysia for providing the facilities to carry out this work.

## COMPETING INTERESTS

Authors have declared that no competing interests exist.

## REFERENCES

- 1. Sources and effects of ionizing radiation, UNSCEAR 2000, United Nation, New York, USA.
- 2. Chibowski S, Polish. Polish J Envir. Stud. 2000;9(4):249-253.
- Myasoedov BF, Pavlotskaya FJ. Measurement of Radioactive Nuclides in the Environment, Analyst. 1989;114: 255-263.
- 4. Kebwaro JM, Rathore IVS, Hasim NO, Mustapha AO. Int. J. Phys. Sci. 2011; 6(13):3105-3110.
- Lieser KH. Radionuclide's in the Geosphere: Sources, Mobility, Reactions in Natural Waters and Interactions with Solids. Radiochimica Acta. 1995;70/71: 355-375.
- 6. Fatima I, Zaidi JH, Arif M, Tahir SNA. Radia. Prot. Dos. 2006;123(2):284-292.
- 7. Ben FA, Hizem N, Chelbi M, Ghedira L. Radiat Prot Dosim. 2005;117: 410-424.
- Kawabata Y, Aparin V, Nagai M, Yamamoto M, Shiraishi K, Katayama Y. Radioanal Nucl Chem. 2008;278(2): 459-462.
- Somlaj J, Horvaath G, KanyaAr B, KovaAcs T, Bodrogi E, KaAvaA si N. J Environ. Radia activ. 2002;62(3):235-240.

- 10. Salih MMI, Pettersson HBL, Lund E. Radia Dosim2002;102(3):249-258.
- 11. Para KS, Kim NB, Kim YS, Lee KY, Choi HW, Yoon YY. J Radioanal and Nucl Chem. 1988;123(2):585-592.
- 12. Rahman ATA, Ramli AT. J Radioanal and Nucl Chem. 2007;273(3):653-657.
- Yusuf NM, Hossain I, Wagiran H. Natural radioactivity in drinking and mineral water in Johor Bahru (Malaysia), Sci. Res. and Essay. 2012;7(9):1070-1075.
- Desidari D, Meli MA, Feduzi L, Roselli C, Rongoni A, Saetta D. <sup>238</sup>U, <sup>234</sup>U, <sup>226</sup>Ra, <sup>210</sup>Po concentrations in bottled mineral waters in Italy and their dose contribution. J. Env. Rad. 2007;94:86-97
- Ajayi OS, Adesida G. Radioactivity in some sachet drinking water samples produced in Nigeria. Iran. J. Radiat. Res. 2009;7(3): 151-158.
- Desideri D, Roselli C, Feduzi L, Meli MA. Radiological characterization of drinking waters in Central. Italy. Microchemical Journal. 2007;87:13-19.
- Fatima I, Zaidi JH, Arif M, Tahir SNA. Measurement of natural radioactivity in bottled drinking water in Pakistan and consequent dose estimates. Rad. Protect. Dos. 2007;123(2):234-240.
- [18] O.S. Ajayi and J.Achuka, Radioactivity in drilled and dug well drinking water of Ogun state Southwestern Nigeria and consequent dose estimates. Rad. Protect. Dos. 2009;135(1):54-63.
- 19. WHO, Guidelines for drinking-water quality, 3rd ed., World Health Organisation, Geneva; 2004.
- 20. ICRP (International Commission on Radiological Protection). Protection of the public in situations of prolonged radiation exposure. ICRP Publication. 2000;82. Ann. ICRP:29 (1-2).

© 2021 Hossain et al.; This is an Open Access article distributed under the terms of the Creative Commons Attribution License (http://creativecommons.org/licenses/by/4.0), which permits unrestricted use, distribution, and reproduction in any medium, provided the original work is properly cited.

Peer-review history: The peer review history for this paper can be accessed here: https://www.sdiarticle4.com/review-history/73946